

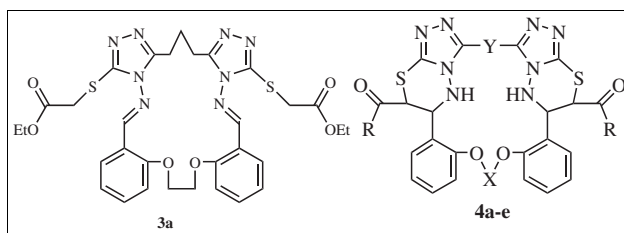
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A practical and regioselective synthetic method for the synthesis of *cis*-diastereomers of bibracchial lariat ethers (BiBLEs) bearing ester and amide groups is reported. The novel BiBLEs **3a** and **4a–e** with neutral side arms were prepared by reaction of the corresponding aza-crown macrocycles **1a–c** with ethylchloroacetate and chloroacetamide. The structures of the new compounds have been confirmed by FTIR, ¹H, ¹³C, DEPT, and MS spectroscopy.

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INTRODUCTION

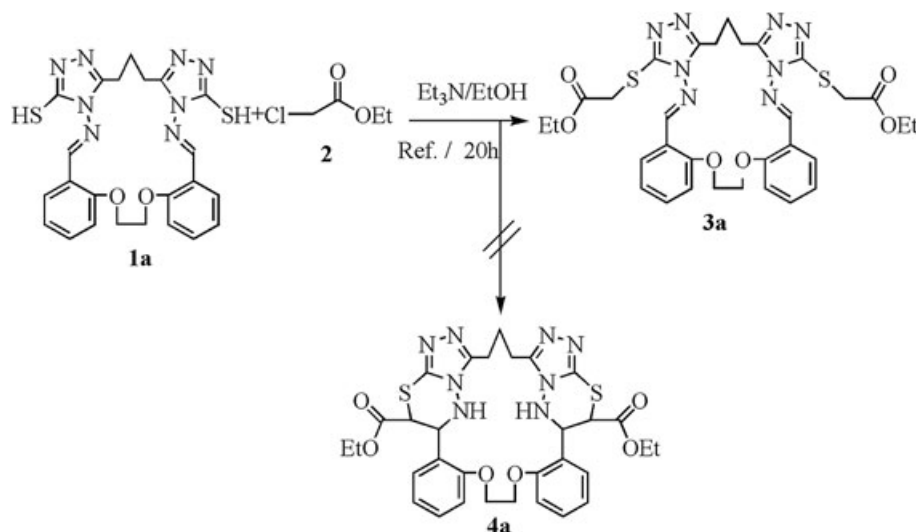
The incorporation of oxygen, nitrogen, and sulfur donor atoms in the structure of macrocycles will significantly affect their complexing properties because of the hard (O, N) and soft (S) character of the donor atoms and the exodentate tendency of the sulfide linkages [1,2]. Other changes involve the insertion of aromatic and/or heterocyclic ring systems into the macrocycles [3–5]; heterocyclic groups provide rigidity and are able in some cases to form complexes *via* their soft donor atoms [6]. The wide interest in the construction of synthetic macrocyclic compounds containing five- and six-membered heterocyclic rings as subunits has led to the preparation of a range of such compounds, which have been shown to possess very interesting properties in a variety of fields [7–10]. Lariat ethers are compounds with a macrocyclic ring and a side arm, which bears a donor group [11–13]. It has been designed as cation complexing agents, which exhibit complexation behavior similar to crown ethers, but with a three-dimensional binding character [14]. Lariat ethers can be divided into two categories depending on the nature of the secondary binding site that with a pendent neutral side arm and lariat ethers with a pendent proton-ionizable arm. Non-ionizable lariat ethers may exhibit an enhanced cation complexation and selectivity when compared with crown ethers without side arms [15].

The intensive development of the lariat crown ether concept has been directed toward the synthesis of several side armed crown ethers, designed for uses ranging from routine (polymer-supported PTC catalysts, separation/

extraction reagents, etc.) to sophisticated applications as redox switches for membrane transport, synthetic cation conducting channels, nucleotide-based molecular boxes, and so on [16,17]. These valuable properties prompted us to synthesize some new series of 18–20 bibracchial lariat Ethers (BiBLEs) containing 1,3,4-thiadiazine-6-carboxylate or 1,3,4-thiadiazine-6-carboxamide.

RESULTS AND DISCUSSION

In continuation of our interest to develop the synthesis of new azathiocrown macrocycles and lariat ether [3,18], we report herein, a simple and efficient method for the regioselective synthesis of novel BiBLEs **4a–e**. In this article, we demonstrate a novel method to introduce 1,2,4-triazolo[3,4-b][1,3,4]thiadiazines rings into macrocycles. The synthesis of BiBLEs containing 1,3,4-thiadiazine-6-carboxylate or 1,3,4-thiadiazine-6-carboxamide has not been reported yet. Aza-crown ether compounds **1a–c** were prepared according to the published method.¹⁵ The functionalities in these aza-crown ethers made them valuable key precursors for the formation of different fused heterocyclic compounds. The available macrocycles **1a–c** encouraged us to study their transformation into the lariat ethers containing ester or amide groups. Thus, the novel lariat compounds **3a** and BiBLEs **4a–e** with neutral side arms were prepared by reaction of corresponding aza-crown macrocycles **1a–c** with ethylchloroacetate and chloroacetamide. The reaction of aza-crown macrocycle with ethylchloroacetate afforded different products depending on the base used.

Scheme 1. Synthesis of lariat ether **3a**.

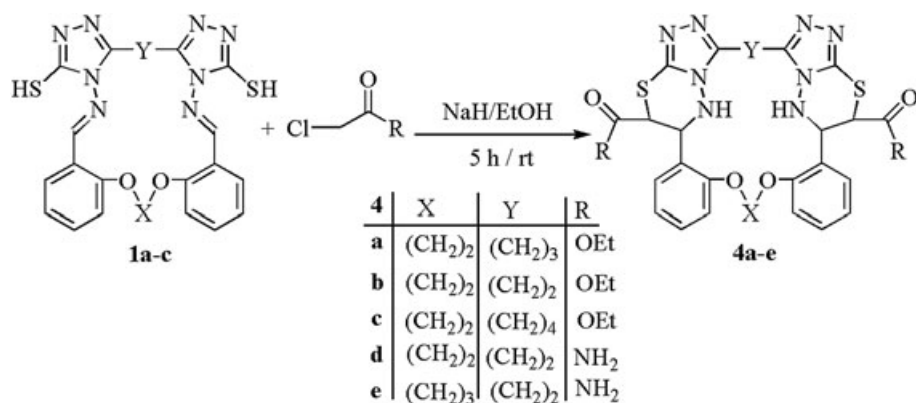
Initially, the reactivity of now available, aza-crown macrocycle with ethylchloroacetate in the present triethylamine as a base in refluxing ethanol was attempted. The reaction of compound **1a** with ethylchloroacetate under reflux conditions did not lead to the formation of BiBLE **4a**. Instead, the reaction gave another product, which could be characterized as the lariat ether **3a** (Schemes 1).

The formation of compound **3a** encouraged us to carry out this reaction in the presence of another base. Thus, stirring of compounds **1a–c** with ethylchloroacetate or chloroacetamide in the presence of sodium hydride for 5 h afforded 58–75% yields of the corresponding novel BiBLEs **4a–e** (Scheme 2).

The reaction proceeds *via* intramolecular cyclocondensation of the active methylene group with the imine group. The expected compounds were obtained in good yields. The reaction of **1a–c** with ethylchloroacetate and chloroacetamide in the presence sodium hydride was

regioselective and afforded only *cis* isomer after ring closure. The isolated compounds **4a–e** were obtained as *cis*-diastereomers. This fact was confirmed by ¹H-NMR data. The stereochemistry of the products was determined from the coupling constant between two vicinal methine protons. In the ¹H-NMR spectra of compounds **4a–e**, the coupling constant (³J_{N-CH, CH-S} ≈ 7.0–8.3 Hz) is typical for the *cis* configuration [19–21].

The IR, ¹H-NMR, and ¹³C-NMR spectra of **4a–e** confirmed the success of the cyclization by the disappearance of the signals corresponding to the SH and CH=N protons and the appearance of signals assigned to the methine and NH protons. The infrared spectra of the aza-crown **1a–c** showed absorptions band, at 2728 cm⁻¹ due to SH groups, which were absent in the IR spectra of compounds **4a–e**. Similarly, the ¹H-NMR spectra of the compounds **1a–c** showed two characteristics absorption (singlet at: δ 8.3 ppm) attributed to the CH=N

Scheme 2. Synthesis of compounds **4a–e**.

groups, and another at: δ 14.2 ppm, assigned to the SH, which were disappeared by the formation of compounds **4a–e**. In addition, the absence of the ^{13}C -NMR and DEPT signals due to the $\text{CH}=\text{N}$ groups and appearance of the aliphatic carbon relative to the thiadiazine ring confirmed the formation of compounds **4a–e**.

CONCLUSIONS

In conclusion, we successfully prepared BiBLEs having pendant groups containing a strong donor group as a supporting ligand at the end of the sidearm.

EXPERIMENTAL

All products were characterized using IR, ^1H -NMR, ^{13}C -NMR spectra and the mass spectral data. All yields refer to isolated products. IR spectra were prepared on a galaxy series FTIR 5000 spectrophotometer using KBr discs. NMR spectra were recorded on Bruker spectrophotometer (300 MHz) in $\text{DMSO}-d_6$ using TMS as an internal standard. Mass spectra were recorded on an Agilent Technology (HP) 5973 Network mass selective detector under electron impact (EI). C, H, N, and S analyses were performed on a Vario EL III elemental analyzer.

General procedure for the synthesis of compounds 3a. A mixture of compound **1a** (0.5 mmol) and triethylamine (3.0 mmol) in ethanol (5 mL) was maintained at reflux for 30 min. A solution of ethylchloroacetate (1.2 mmol) in ethanol (5 mL) was added, and the mixture was refluxed for 20 h. The reaction was monitored by TLC. After the completion of the reaction, the solvent was removed under reduced pressure. The crude solid was dissolved in CHCl_3 , and the solution was washed with water. The organic layer was separated and dried over anhydrous sodium sulfate. The organic solvent was removed under reduced pressure. The residue was crystallized from ethylacetate to give the pure compound **3a**.

Ethyl-2-(9-[2ethoxy-2-oxoethylsulfanyl]-5,6,18,19-tetrahydro-4H-dibenzo[e,r]di[1,2,4]triazolol[4,3-i:3,4-n][1,4,8,9,15,16]dioxatetraacyclononadecin-1-yl)sulfanyl)acetate 3a. 86% yield, IR (KBr): ν (cm^{-1}): 3062 (aromatic CH stretch.), 2937 (aliphatic CH stretch.) 1735 (C=O), 1601 (C=N), 1250, 1161, ^1H -NMR (300 MHz, $\text{DMSO}-d_6$): δ 1.09 (t, 6H, 2CH_3 , $J = 7.0$ Hz), 2.05 (t, 2H, CH_2 , $J = 6.2$ Hz), 2.77 (t, 4H, 2CH_2 , $J = 6.1$ Hz), 3.85 (s, 4H, 2SCH_2), 3.98 (q, 4H, 2OCH_2 , $J = 7.0$ Hz), 4.52 (s, 4H, 2OCH_2), 7.14 (t, 2H, $\text{H}_{\text{arom.}}$, $J = 7.0$ Hz), 7.27 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 8.3$ Hz), 7.61 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 7.8$ Hz), 7.75 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 7.6$ Hz), 8.78 (s, 2H, $2\text{CH}=\text{N}$), ^{13}C -NMR (75 MHz, $\text{DMSO}-d_6$): δ 14.3, 23.2, 24.9, 34.7, 61.6, 68.3, 114.5, 120.4, 121.9, 127.3, 135.5, 145.2, 152.6, 159.5, 162.0, 168.4. *Anal.* Calcd. for: $\text{C}_{31}\text{H}_{34}\text{N}_8\text{O}_6\text{S}_2$: C, 54.85; H, 5.05; N, 16.51; S, 9.45 Found: C, 54.57; H, 4.99; N, 16.31; S, 9.26.

General procedure for the synthesis of compounds 4a–e. Sodium hydride (2.5 mmol) was added to a solution of compounds **1a–c** (0.5 mmol) in absolute ethanol (10 mL) at room temperature. Salt formation was allowed to proceed at room temperature for 10 min and ethylchloroacetate or chloroacetamide (1.1 mmol) was added and the solution stirred for 5 h at room temperature. After the completion of

the reaction, the solvent was removed under vacuum and extracted with ethylacetate; the organic layer was washed with water (3×10 mL), dried (Na_2SO_4), and evaporated under vacuum. The residue was crystallized from ethylacetate and petroleum ether to give compounds **4a–e**.

Diethyl-8,11-dioxa-32,36-dithia-19,20,22,23,29,30,34,35-octaazaheptacyclo[26.5.2.2 18,21 .0 2,7 .0 12,17 .0 20,24 .0 31,35]heptatriaconta-2,4,6,12(17),13,15,21,23,28,30-decaene-33,37-dicarboxylate (4a). 70% yield, IR (KBr): ν (cm^{-1}): 3250 (NH stretch.) 3064 (aromatic CH stretch.), 2955 (aliphatic CH stretch.) 1732 (C=O), 1610 (C=N), 1244, 1165, ^1H -NMR (300 MHz, $\text{DMSO}-d_6$): δ 0.97 (t, 6H, 2CH_3 , $J = 7.0$ Hz), 2.00 (br, 2H, CH_2), 2.76 (br, 4H, 2CH_2), 3.99 (q, 4H, 2CH_2 , $J = 7.1$ Hz), 4.35 (d, 2H, $2\text{S}-\text{CH}$, $J = 7.8$ Hz), 4.44 (d, 2H, $2\text{N}-\text{CH}$, $J = 8.0$ Hz), 4.78 (s, 4H, 2OCH_2), 6.79 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 7.1$ Hz), 7.02 (t, 2H, $\text{H}_{\text{arom.}}$, $J = 7.0$ Hz), 7.15 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 8.2$ Hz), 7.30 (s, 2H, $2\text{NH}\cdot\text{D}_2\text{O}$ exchange), 7.34 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 5.0$ Hz), ^{13}C -NMR (75 MHz, $\text{DMSO}-d_6$): δ 14.0, 23.1, 25.2, 40.3, 44.5, 61.8, 68.3, 113.4, 121.5, 124.1, 128.1, 130.8, 141.2, 152.5, 156.7, 168.9, DEPT: δ 14.0 (CH_3) 23.1 (CH_2), 25.2 (CH_2), 40.3 (CH), 44.5 (CH), 61.8 (CH_2), 68.3 (CH_2), 113.4 (CH), 121.5 (CH), 128.1 (CH), 130.8 (CH), $m/z = 678$ (M^+), 663, 648, 634, 588, 534, 324, 264, 146 (base peak), 120, 91. *Anal.* Calcd. for: $\text{C}_{31}\text{H}_{34}\text{N}_8\text{O}_6\text{S}_2$: C, 54.85; H, 5.05; N, 16.51; S, 9.45 Found: C, 54.51; H, 4.97; N, 16.33; S, 9.22.

Diethyl-8,11-dioxa-31,35-dithia-19,20,22,23,28,29,33,34-octaazaheptacyclo[25.5.2.2 18,21 .0 2,7 .0 12,17 .0 20,24 .0 30,34]hexatriaconta-2,4,6,12(17),13,15,21,23,27,29-decaene-32,36-dicarboxylate (4b). 70% yield, IR (KBr): ν (cm^{-1}): 3256 (NH stretch.) 3070 (aromatic CH stretch.), 2924 (aliphatic CH stretch.) 1730 (C=O), 1601 (C=N), 1248, 1161, ^1H -NMR (300 MHz, $\text{DMSO}-d_6$): δ 0.97 (t, 6H, 2CH_3 , $J = 6.8$ Hz), 2.93 (s, 4H, 2CH_2), 3.97 (q, 4H, 2CH_2 , $J = 6.5$ Hz), 4.30 (d, 2H, $2\text{S}-\text{CH}$, $J = 8.1$ Hz), 4.46 (d, 2H, $2\text{N}-\text{CH}$, $J = 8.3$ Hz), 4.77 (s, 4H, 2OCH_2), 6.72 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 7.2$ Hz), 7.00 (t, 2H, $\text{H}_{\text{arom.}}$, $J = 7.2$ Hz), 7.13 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 8.0$ Hz), 7.33 (s, 2H, $2\text{NH}\cdot\text{D}_2\text{O}$ exchange), 7.36 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 5.1$ Hz), ^{13}C -NMR (75 MHz, $\text{DMSO}-d_6$): δ 14.0, 21.5, 40.1, 44.2, 62.1, 67.8, 113.0, 121.3, 124.6, 128.0, 130.6, 141.8, 152.9, 156.3, 168.8, DEPT: δ 14.0 (CH_3) 21.5 (CH_2), 40.2 (CH), 44.2 (CH), 62.2 (CH_2), 67.8 (CH_2), 113.0 (CH), 121.3 (CH), 128.0 (CH), 130.6 (CH), $m/z = 664$ (M^+), 649, 634, 621, 606, 591, 577, 574, 551, 284, 264, 146 (base peak), 120, 91. *Anal.* Calcd. for: $\text{C}_{30}\text{H}_{32}\text{N}_8\text{O}_6\text{S}_2$: C, 54.20; H, 4.85; N, 16.86; S, 9.65. Found: C, 53.95; H, 4.75; N, 16.53; S, 9.46.

Diethyl-8,11-dioxa-33,37-dithia-19,20,22,23,30,31,35,36-octaazaheptacyclo[27.5.2.2 18,21 .0 2,7 .0 12,17 .0 20,24 .0 32,36]octatriaconta-2,4,6,12(17),13,15,21,23,28,31-decaene-34,38-dicarboxylate (4c). 58% yield, IR (KBr): ν (cm^{-1}): 3261 (NH stretch.) 3054 (aromatic CH stretch.), 2933, 2852 (aliphatic CH stretch.) 1732 (C=O), 1600 (C=N), 1240, 1161, ^1H -NMR (300 MHz, $\text{DMSO}-d_6$): δ 1.12 (br, 6H, 2CH_3), 1.54 (br, 4H, 2CH_2), 2.72 (s, 4H, 2CH_2), 3.68 (q, 4H, 2CH_2 , $J = 6.7$ Hz), 4.38 (d, 2H, $2\text{S}-\text{CH}$, $J = 8.1$ Hz), 4.49 (d, 2H, $2\text{N}-\text{CH}$, $J = 8.0$ Hz), 4.65 (s, 4H, 2OCH_2), 6.81 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 7.5$ Hz), 7.04 (t, 2H, $\text{H}_{\text{arom.}}$, $J = 7.0$ Hz), 7.11 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 6.8$ Hz), 7.25 (s, 2H, $2\text{NH}\cdot\text{D}_2\text{O}$ exchange), 7.30 (d, 2H, $\text{H}_{\text{arom.}}$, $J = 5.7$ Hz), ^{13}C -NMR (75 MHz, $\text{DMSO}-d_6$): δ 14.0, 23.0, 23.7, 40.6, 44.0, 62.4, 68.8, 113.1, 121.0, 124.2, 128.1, 130.0, 141.5, 153.0, 156.1, 168.8, DEPT: δ 14.0 (CH_3) 23.0 (CH_2), 23.7 (CH_2), 40.6 (CH), 44.0 (CH), 62.4 (CH_2), 68.8 (CH_2), 113.1 (CH), 121.0 (CH), 128.1 (CH), 130.0; $m/z = 692$ (M^+), 677, 634, 604, 284, 264, 146

(base peak), 120, 91. *Anal.* Calcd. for: C₃₂H₃₆N₈O₆S₂: C, 55.48; H, 5.24; N, 16.17; S, 9.26. Found: C, 55.13; H, 5.03; N, 15.98; S, 9.00.

8,11-Dioxa-31,35-dithia-19,20,22,23,28,29,33,34-octaazaheptacyclo[25.5.2.2^{18,21}.0^{2,7}.0^{12,17}.0^{20,24}.0^{30,34}]hexatriaconta-2,4,6,12(17), 13,15,21,23,27,29-decaene-32,36-dicarboxamide (4d). 75% yield, IR (KBr): ν (cm⁻¹): 3315, 3184 (NH₂ stretch.) 3060 (aromatic CH stretch.), 2951 (aliphatic CH stretch.) 1689 (C=O), 1251, 1114, ¹H-NMR (300 MHz, DMSO-d₆): δ 2.90 (s, 4H, 2CH₂), 4.22 (d, 2H, S—CH, *J* = 7.0 Hz), 4.43 (d, 2H, N—CH, *J* = 7.1 Hz), 4.61 (br, 2H, OCH₂), 4.77 (br, 2H, OCH₂), 6.62 (s, 2H, NH₂, D₂O exchange), 7.01 (t, 2H, H_{arom.}, *J* = 6.4 Hz), 7.12 (d, 2H, H_{arom.}, *J* = 7.5 Hz), 7.23 (s, 2H, 2 NH·D₂O exchange), 7.34 (t, 2H, H_{arom.}, *J* = 6.8 Hz), 7.43 (d, 2H, H_{arom.}, *J* = 5.7 Hz), 7.86 (s, 2H, NH₂, D₂O exchange), ¹³C-NMR (75 MHz, DMSO-d₆): δ 21.7, 40.3, 43.7, 68.0, 113.1, 121.1, 124.8, 127.8, 130.5, 142.4, 152.8, 156.7, 169.6, DEPT: δ 21.7 (CH₂), 40.3 (CH), 43.7 (CH), 68.0 (CH₂), 113.0 (CH), 121.1 (CH), 127.8 (CH), 130.5 (CH), MS (EI): *m/z* = 606 (M⁺), 577, 551, 395, 365, 339, 264, 146, 119, 91, 57, 43 (base peak). *Anal.* Calcd. for: C₂₆H₂₆N₁₀O₆S₂: C, 51.47; H, 4.32; N, 23.09; S, 10.57. Found: C, 51.23; H, 4.23; N, 22.86; S, 10.41.

8,12-Dioxa-32,36-dithia-20,21,23,24,29,30,34,35 octaazaheptacyclo[26.5.2.2^{19,22}.0^{2,7}.0^{13,18}.0^{21,25}.0^{31,35}]heptatriaconta-2,4,6, 13(18),14,16,22,24,28,30-decaene-33,37-dicarboxamide (4e). 65% yield, IR (KBr): ν (cm⁻¹): 3330, 3176 (NH₂ stretch.) 3054 (aromatic CH stretch.), 2833–2941 (aliphatic CH stretch.) 1676 (C=O), 1242, ¹H-NMR (300 MHz, DMSO-d₆): δ 2.26 (br, 2H, CH₂), 3.00 (s, 4H, 2CH₂), 4.19 (d, 2H, S—CH, *J* = 7.3 Hz), 4.49 (d, 2H, N—CH, *J* = 7.3 Hz), 4.62 (br, 2H, OCH₂), 4.79 (br, 2H, OCH₂), 6.70 (s, 2H, NH₂, D₂O exchange), 7.01 (t, 2H, H_{arom.}, *J* = 6.5 Hz), 7.14 (d, 2H, H_{arom.}, *J* = 8.1 Hz), 7.26 (s, 2H, 2 NH·D₂O exchange), 7.38 (t, 2H, H_{arom.}, *J* = 6.7 Hz), 7.56 (d, 2H, H_{arom.}, *J* = 5.9 Hz), 7.78 (s, 2H, NH₂, D₂O exchange), ¹³C-NMR (75 MHz, DMSO-d₆): δ 21.3, 29.1, 40.1, 43.8, 65.3, 113.2, 120.9, 124.6, 128.0, 130.31, 142.46, 152.9, 156.7, 169.5, DEPT: δ 22.1 (CH₂), 22.8 (CH₂), 40.1 (CH), 43.8 (CH), 67.5 (CH₂), 112.9 (CH), 121.5 (CH), 128.0 (CH), 130.3 (CH), MS (EI): *m/z* = 620 (M⁺), 590, 576, 532, 278, 146, 119, 57, 43 (base peak). *Anal.* Calcd. for: C₂₇H₂₈N₁₀O₆S₂: C, 52.25; H, 4.55; N, 22.57; S, 10.33. Found: C, 52.04; H, 4.39; N, 22.41; S, 10.18.

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REFERENCES AND NOTES

- [1] Elwahy, A. H. M.; Abbas, A. A.; Ibrahim, Y. A. *J Chem Res* (s) 1996, 182.
- [2] Richard, E. D.; Milton, D. G. *J Am Chem Soc* 1976, 98, 762.
- [3] Foroughifar, N.; Mobinikhaldei, A.; Ebrahimi, S.; Moghanian, H.; Bodaghi Fard, M. A.; Kalhor, M. *Tetrahedron Lett* 2009, 50, 836.
- [4] Weber, E.; Kohler, H. J.; Reuter, H. J. *J Org Chem* 1991, 56, 1236.
- [5] Muathen, H. A.; Aloweiny, N. A. M.; Elwahy, A. H. M. *J Heterocycl Chem* 2009, 46, 656.
- [6] Newkome, G. R.; Sauer, J. D.; Robber, J. M.; Hager, D. C. *Chem Rev* 1977, 77, 513.
- [7] Bradshaw, J. S.; Huzzthy, P.; McDaniel, C. W.; Zhu, C. Y.; Dalley, N. K.; Izatt, R. M. *J Org Chem* 1990, 55, 3129.
- [8] Bradshaw, J. S.; Thompson, P. K.; Izatt, R. M.; Morin, F. G.; Grant, D. M. *J Heterocycl Chem* 1984, 21, 897.
- [9] Ibrahim, Y. A.; Elwahy, A. H. M.; Abbas, A. A.; Kassab, R. M. *J Chem Res* (s) 1999, 522.
- [10] Elwahy, A. H. M.; Masaret, G. S. *J Heterocycl Chem* 2007, 44, 1475.
- [11] Sharghi, H.; Khalifeh, R.; Salimi Beni, A.R. *J Iranian Chem Soc* 2010, 7, 275.
- [12] Elwahy, A. H. M.; Abbas, A. A. *J Heterocycl Chem* 2008, 45, 1.
- [13] Abbas, A. A.; Elwahy, A. H. M. *J Heterocycl Chem* 2009, 46, 1035.
- [14] Weber, E.; Vogtle, F. In *Crown-Type Compounds: An Introductory Overview. Host Guest Complex Chemistry, Part 1/Topics in Current Organic Chemistry Series*; Vogtle, F., Ed.; Springer: New York, 1981; pp 14.
- [15] Kolthoff, I. M. *Anal Chem* 1979, 51, 1R.
- [16] Gokel, G. W.; Dishong, D. M.; Diamond, C. J. *J Chem Soc Chem Commun* 1980, 1053.
- [17] Schultz, R. A.; White, B. D.; Dishong, D. M.; Arnold, K. A.; Gokel, G. W. *J Am Chem Soc* 1985, 107, 6659.
- [18] Foroughifar, N.; Mobinikhaldei, A.; Ebrahimi, S. *Synthesis* 2009, 15, 2557.
- [19] Foroughifar, N.; Mobinikhaldei, A.; Ebrahimi, S. *Synth Commun* 2010, 40, 2421.
- [20] Elwahy, A. H. M.; Abbas, A. A.; Ahmed, A. A. M. *J Heterocycl Chem* 2005, 42, 93.
- [21] Ibrahim, Y. A.; Elwahy, A. H. M.; El-Fiky, A. E. M. *Heteroatom Chem* 1994, 5, 321.